Taxamycin Studies: CrCl₂/NiCl₂-Mediated Cyclizations and Unusual SeO₂ Oxidations

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The enediyne family of anticancer antibiotics continues to be a topic of considerable current interest. This is a consequence of their novel mechanism of DNA cleavage involving cycloaromatization to benzyne diradicals and the synthetic challenge presented by the natural products themselves. This has also stimulated the design and synthesis of simpler analogues.¹ Hybrid structures that mimic esperamicin² and calicheamicin³ include incorporation of part of the taxane framework (taxamycins),⁴ the steroid nucleus (estramycins),⁵ β -lactams (lactendiynes),⁶ and enediynes linked to diethylstilbestrol.⁷ In a continuation of our studies of the taxamycins 1, we wish to report a more direct route to the taxamycin-12 compound 15 (eight fewer steps), the synthesis of the taxamycin-11 compound 16, a study of CrCl₂/NiCl₂-mediated cyclizations, and an unusual SeO₂ oxidation of the α -acetylenic ethers 17 via an allene intermediate that may provide an alternative cycloaromatization route in appropriate systems.



Aldehydes **2** and **4**⁸ and enediyne **3** were prepared as previously described.⁴ Condensation of the aldehyde **2** with the lithium acetylide derived from **3** afforded an 83% yield of the secondary alcohol diastereomer **5** (ratio

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Several methods have been examined for the intramolecular ring closure of medium-sized cyclic enediynes, particularly for the 10-membered rings that mimic calicheamicin and esperamicin. These include ring contraction of a cyclic ether by [2,3]-Wittig rearrangement,¹⁰ Nicholas-type cyclization of η^2 -dicobalthexacarbonyl complexed propargylic cations with Lewis acid,¹¹ acetylenic anion–aldehyde condensations,¹² and fluoride-induced desilylation–condensation.¹³ Our preferred method is the extension of the chromium(II)-mediated Nozaki– Kishi coupling¹⁴ to intramolecular cyclizations of iodoacetylenes with aldehydes developed by Crevisy and Beau.¹⁵

In order to establish a reliable protocol, a number of experiments were conducted (table in the Supporting Information). Reactions f and l in Scheme 1 in THF,¹⁶ using the CrCl₂(THF) complex,^{17,18} represent the best cases. (DMF, useful for intermolecular examples, etc., failed.) The quantity of NiCl₂ was determined experimentally as the literature indicated that for some 10-membered ring systems nickel chloride was not required.¹⁶ In other cases,¹⁹ low yields were obtained with catalytic quantities and stoichiometric amounts were better.

Base-induced cyclization of **14** ($\mathbf{R} = \mathbf{H}$) with 2 equiv of potassium hexamethyldisilazide at -78 °C also provided the cyclized material **16** in variable yields of 15–25%.

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^a Reagents: (a) *n*-BuLi, THF, -78 to +21 °C, 15 h, 83% (15:1); (b) MOMCl, (*i*-Pr)₂NEt, CH₂Cl₂, 0-21 °C, 15 h, 85%; (c) TBAF, THF, -78 to +21 °C, 3 h, 99%; (d) I₂, morpholine, C₆H₆, 60–70 °C, 3 h, 83%; (e) Dess-Martin periodinane, CH₂Cl₂, 21 °C, 30 min, 92%; (f) CrCl₂ (THF) (6 equiv), NiCl₂ (1.6 equiv), THF, 30 min, 40%; (g) *n*-BuLi, -78 to +21 °C, 15 h, 86% (4:1); (h) NaH, MeI, THF, 0-21 °C, 5 h, 89%; (i) TBAF, THF, -78 to +21 °C, 1 h, 91%; (j) I₂ morpholine, C₆H₆, 60–70 °C, 3 h, 75%l (k) 3:1 THF/4% aqueous HCl, 3 h, 91%; (l) CrCl₂(THF) (8 equivf), NiCl₂ (0.06 equiv), THF, 21 °C, 45 min, 37%.

Anhydrous fluoride sources such as tetrabutylammonium difluorotriphenylstannate²⁰ caused desilylation and no cyclization of **14** (R = TIPS). The new secondary alcohols in **15** and **16** were syn relative to the C₂ ether groups, a stereochemical result that was consistent with related intramolecular cases in which the same geometric preference was observed for closure to 10-membered rings.^{12d}

It was anticipated, by analogy with taxoid systems, that allylic oxidation would introduce the desired oxygen at C_{13} (taxane numbering)^{9c,21} for attachment of the appropriate side chain. However, oxidants such as CrO_3 or PCC resulted only in recovery of starting material.



Treatment of the acetate 17a with selenium dioxide in dioxane at 60-70 °C afforded exclusively the C2 ketone 18a in 89% yield. The strain and lack of orbital overlap in this ring system is reflected in the carbonyl IR band at 1740 cm^{-1} . Initially it appeared that the acidity of the reaction had resulted in hydrolysis of the MOM ether followed by oxidation of the resulting alcohol. However, this was not the case, as a parallel oxidation of the methyl ether 17b with excess SeO_2 also gave the corresponding ketone 18b in a yield of 77% (NMR COSY). Consistent with the accepted mechanism of selenium dioxide oxidations,22 these reactions likely proceeded via an ene reaction to form the allene intermediate 19 followed by sigmatropic rearrangement to an acetal and hydrolysis to the ketones upon aqueous workup. The double bond in the A ring is hindered by both the enediyne bridge on one face and by the sterically demanding gem dimethyl group on the other. These features apparently inhibit the approach of the SeO₂, and thus, the rearrangement pathway is observed. In principle, the generation of the allene should facilitate the cycloaromatization (Meyers cyclization), but this was not observed due to the presence of the bridgehead olefin. However, this oxidation sequence may have potential as a trigger mechanism in other cases.

In summary, general methods have been developed for the synthesis of these challenging bridged enediyne systems using chromium/nickel-mediated coupling. These cyclizations are difficult due to the strained rings produced in addition to the steric hindrance provided by the geminal methyl groups. Investigations in progress have demonstrated that in less demanding cases the yields are improved under the conditions developed above. These results will be reported in due course.

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Supporting Information Available: Experimental procedures and spectral data (24 pages).

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